

Yong-Qiang Yu, Yue Wang,
Pei-Zhou Ni and Tao Lu*College of Basic Science, China Pharmaceutical
University, Nanjing 210009, People's Republic
of China

Correspondence e-mail: lut163@163.com

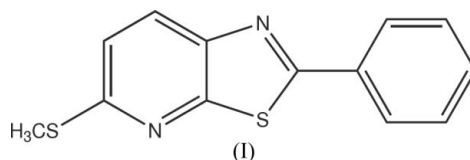
Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.006$ Å
 R factor = 0.052
 wR factor = 0.201
Data-to-parameter ratio = 15.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.6-(Methylsulfonyl)-2-phenylthiazolo[5,4-*b*]pyridine

The molecule of the title compound, $\text{C}_{13}\text{H}_{10}\text{N}_2\text{S}_2$, is approximately planar, the thiazolopyridine ring system being slightly twisted from the phenyl ring. The molecules are stacked along the b axis through π - π interactions.

Comment

Thiazolopyridine derivatives have been widely studied because of their analgesic, antipyretic and anti-inflammatory properties (Couture & Grandclaude, 1985, 1987). Most of them have a methyl or methoxy substituent attached to the pyridine ring of the thiazolopyridine unit (Kaigorodova, Osipova *et al.*, 2004; Kaigorodova, Vasilin *et al.*, 2004). As part of an investigation of the antitumor properties of the thiazolopyridine analogs, we report the synthesis and structure of the title compound, (I), containing a methylsulfonyl substituent attached to the pyridine ring.



The molecular structure is approximately planar (Fig. 1), the thiazolopyridine ring system and the phenyl ring being slightly twisted by $9.685(12)^\circ$, which is comparable to the corresponding angles reported in 5-dimethylamino-2-phenyl-4,7-dioxobenzothiazole (Lyon *et al.*, 1999) and 5-benzyl-2-phenyl-4-tolylthiazole (Bart *et al.*, 1978). The bond distances of $\text{S1}-\text{C7}$ and $\text{S1}-\text{C12}$ in (I) (Table 1) are a little longer than those in other thiazolo derivatives (Kasim & Yamin, 2005; Cui *et al.*, 2005). The methylsulfonyl group is slightly twisted from the pyridine ring plane with a $\text{C13}-\text{S2}-\text{C11}-\text{N2}$ torsion angle of $6.0(4)^\circ$. The $\text{S2}-\text{C11}$ bond is longer than the reported values of $1.713(2)-1.765(2)$ Å (Low *et al.*, 1996*a,b*; Yu *et al.*, 1996), while the $\text{S2}-\text{C13}$ bond is a little shorter than the corresponding values of $1.795(2)$ Å (Lipinska, 2005) and $1.823(2)$ Å (Low *et al.*, 1996*b*). In the crystal structure, the molecules are stacked along the b axis, forming a column in which the short intermolecular contact between the thiazole $\text{S1}/\text{C7}/\text{N1}/\text{C8}/\text{C12}$ and phenyl $\text{C1}-\text{C6}$ rings indicates the existence of a π - π stacking interaction [$\text{Cg1} \cdots \text{Cg2}^i = 3.759(2)$ Å, where Cg1 and Cg2 are the centroids of the thiazole and phenyl rings, respectively; symmetry code: (i) $x, -1 + y, z$].

Experimental

To an anhydrous xylene solution (10 ml) of methyl methylthiomethyl sulfoxide (8 mmol, 0.1 g) was added sodium hydride (8 mmol,

Received 18 November 2006
Accepted 19 January 2007

0.032 g) under a nitrogen atmosphere. The mixture was heated to 337 K and 2-phenyl-6-chlorothiazolo[5,4-*b*]pyridine (0.2 g) was added. After refluxing the solution for 8 h, the reaction mixture was poured into water (50 ml). The product was extracted with ethyl acetate and the organic layer was dried with magnesium sulfate. The solvent was removed under reduced pressure and the residual product was chromatographed to give pure (I) (yield 70%, 0.14 g). Single crystals suitable for X-ray diffraction were obtained from an *n*-hexane solution at room temperature.

Crystal data

$C_{13}H_{10}N_2S_2$	$Z = 4$
$M_r = 258.35$	$D_x = 1.433 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 19.467 (4) \text{ \AA}$	$\mu = 0.42 \text{ mm}^{-1}$
$b = 5.063 (1) \text{ \AA}$	$T = 293 (2) \text{ K}$
$c = 12.152 (2) \text{ \AA}$	Plate, colorless
$\beta = 90.09 (3)^\circ$	$0.40 \times 0.30 \times 0.10 \text{ mm}$
$V = 1197.7 (4) \text{ \AA}^3$	

Data collection

Enraf–Nonius CAD-4 diffractometer	2346 independent reflections
$\omega/2\theta$ scans	1608 reflections with $I > 2\sigma(I)$
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	$R_{\text{int}} = 0.021$
$T_{\text{min}} = 0.850$, $T_{\text{max}} = 0.959$	$\theta_{\text{max}} = 26.0^\circ$
2417 measured reflections	3 standard reflections every 200 reflections
	intensity decay: none

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.12P)^2 + 0.55P]$
$R[F^2 > 2\sigma(F^2)] = 0.052$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.202$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.05$	$\Delta\rho_{\text{max}} = 0.37 \text{ e \AA}^{-3}$
2346 reflections	$\Delta\rho_{\text{min}} = -0.41 \text{ e \AA}^{-3}$
154 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (\AA , $^\circ$).

S1–C12	1.740 (4)	S2–C11	1.763 (4)
S1–C7	1.750 (4)	S2–C13	1.789 (5)
C12–S1–C7	88.88 (19)	C11–S2–C13	102.4 (2)

All H atoms were located in a difference Fourier map, and then they were placed in idealized positions ($C-H = 0.93-0.96 \text{ \AA}$) and treated as riding, with $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ or $1.5U_{\text{eq}}(\text{methyl } C)$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms &

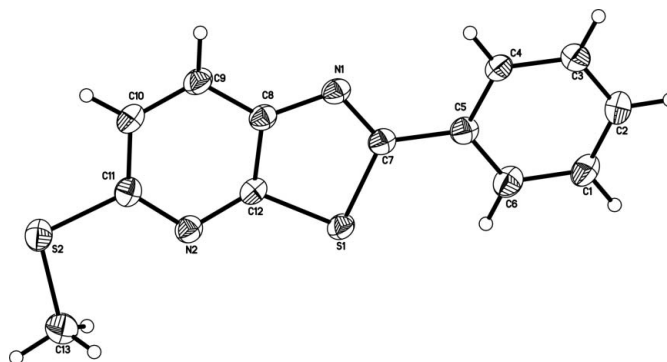


Figure 1

The molecular structure of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

The laboratory of Professor Nobuo Okabe (Kinki University, Japan) is thanked for assistance.

References

- Bart, J. C. J., Bassi, I. W., Benedicenti, C., Calcaterra, M. & Intrito, R. (1978). *Acta Cryst.* **B34**, 3639–3644.
- Couture, A. & Grandclaudeon, P. (1985). *Synthesis*, pp. 533–535.
- Couture, A. & Grandclaudeon, P. (1987). *J. Heterocycl. Chem.* **24**, 1765–1769.
- Cui, Z.-P., Liu, J.-C. & He, H.-W. (2005). *Acta Cryst.* **E61**, o3642–o3644.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.
- Kaigorodova, E. A., Osipova, A. A., Konyushkin, L. D. & Krapivin, G. D. (2004). *Russ. Chem. Bull. (Engl. Transl.)*, **53**, 853–859.
- Kaigorodova, E. A., Vasilin, V. K., Lipunov, M. M., Zavodnik, V. E. & Krapivin, G. D. (2004). *Chem. Heterocycl. Compd.* **40**, 1600–1608.
- Kasim, N. A. M. & Yamin, B. M. (2005). *Acta Cryst.* **E61**, o3794–o3795.
- Lipinska, T. (2005). *Tetrahedron*, **61**, 8148–8158.
- Low, J. N., Ferguson, G., Cobo, J., Melguizo, M., Noguera, M. & Sánchez, A. (1996a). *Acta Cryst.* **C52**, 143–145.
- Low, J. N., Ferguson, G., Cobo, J., Melguizo, M., Noguera, M. & Sánchez, A. (1996b). *Acta Cryst.* **C52**, 145–148.
- Lyon, M. A., Lawrence, S., Williams, D. J. & Jackson, Y. A. (1999). *J. Chem. Soc. Perkin Trans. 1*, pp. 437–442.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Yu, P., Zhuang, B., Huang, L. & Pan, B. (1996). *Acta Cryst.* **C52**, 630–632.